## Quantum Mechanics II: HW 1

- 1. Analytical Methods: Problems 4.79, 4.80, 4.82, 4.96, 4.97, 4.98.
- 2. QM Notes: Problems 9.1 and 9.2.
- 3. Weinberg 4.3, 4.4, 4.5, 4.6, 4.7, 4.9
- 4. If  $Y_{\ell}^{m}$  is a spherical harmonic and if  $|\vec{X}|^{\ell}Y_{\ell}^{m}(\hat{X})$  is viewed as an operator built out of the position operator  $\vec{X}$ , use the Wigner-Eckart theorem to evaluate the following solid angle integral on the 2-sphere in terms of Clebsch-Gordan coefficients:

$$\int_{\mathbb{S}^2} \overline{Y_{\ell_1}^{m_1}(\widehat{x})} Y_{\ell_2}^{m_2}(\widehat{x}) Y_{\ell_3}^{m_3}(\widehat{x}) \mathrm{d}\Omega_{\widehat{x}}.$$
 (0.0.1)

5. Consider two spin-half particles in a bound system, such that the total Hamiltonian can be written in terms of their reduced mass  $\mu$ ; relative coordinate  $\vec{X}$ ; relative orbital angular momentum  $\vec{L}$ ; their individual and total spin operators  $\vec{S'}$ ,  $\vec{S''}$  and  $\vec{S} \equiv \vec{S'} + \vec{S''}$ ; as well as the total angular momentum  $\vec{J} \equiv \vec{L} + \vec{S}$ :

$$H\psi = -\frac{1}{2\mu} \left\{ \frac{1}{r^2} \partial_r \left( r^2 \partial_r \psi \right) - \frac{\vec{L}^2}{r^2} \psi \right\} + \left\{ V_0(r) + V_1(r) \left( \vec{S}' \cdot \vec{S}'' \right) + V_3(r) \left( \vec{L} \cdot \vec{S} \right) \right\} \psi,$$
  

$$r \equiv |\vec{x}|.$$
(0.0.2)

Use the separation-of-variables technique – i.e., assume the wave function is a function of  $\vec{x}$  multiplied by some spin-dependent state – and write down the ordinary differential equation for the energy eigenstate with total angular momentum j, total orbital momentum  $\ell$  and total spin  $\sigma$ .<sup>1</sup> Hint: First explain why

$$\vec{S}' \cdot \vec{S}'' = \frac{1}{2} \left( \vec{S}^2 - \vec{S}'^2 - \vec{S}''^2 \right), \qquad (0.0.3)$$

$$\vec{L} \cdot \vec{S} = \frac{1}{2} \left( \vec{J}^2 - \vec{L}^2 - \vec{S}^2 \right).$$
(0.0.4)

<sup>&</sup>lt;sup>1</sup>This problem can be found in at least one quantum mechanics text – which shall be revealed in the solutions – so make sure you explain your steps carefully!